



KUGESI

DISTRIBUTORS OF REFRIGERANT

MATERIAL DATA SAFETY SHEET

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Date: 1st September 2008

1) Product and company identification

PRODUCT NAME: Refrigerant Gas 22
SUPPLIER DETAILS: Kugesi Refrigerants (Pty) Ltd
8 Benetton Street
Racing Park
Killarney
Cape Town, ZA
Tel: +27 21 556 2652
Fax: +27 21 556 2717

2) Composition/Information on ingredients

Chemical nature: Halogenated Hydrocarbon
Hazardous Constituents: Chlorodifluoromethane

3) Hazards identification

Adverse human health effects: Liquefied gas, contact of liquid may cause frostbite and injury to the Cornea. High exposures may cause an abnormal heart rhythm and prove suddenly fatal. Very high atmospheric concentrations may cause anaesthetic effects and asphyxiation. Liquid splashes may cause freeze burn to the skin and eyes.

Physical and Chemical hazards: Heating will cause a rise in pressure and risk of bursting. On combustion toxic gasses are released.

4) First Aid Measures

Inhalation: Move the affected person away from the contaminated area and into fresh air. Keep the person warm and at rest. Administer oxygen if necessary. If breathing stops give artificial respiration. In the event of cardiac arrest apply external cardiac massage. Call a doctor immediately.

Skin Contact: Contact of liquid with Skin: Immediately rinse plenty of water to thaw the affected area. Immediately remove contaminated clothing or footwear. Clothing may adhere to the skin, if it sticks do not pull it off. Cover affected area with sterile dressing. If irritation or blistering occur seek immediate medical attention.

Eye Contact: Contact of liquid with eyes: rinse with water whilst keeping the eyes wide open irrigate for at least 10 minutes. Consult an eye specialist immediately.

Ingestion: This is an unlikely route of exposure. If exposure takes place do not induce vomiting. Providing the patient is conscious, wash out the mouth with water and give 200-300ml of water to drink. Obtain immediate medical attention as ingestion will cause freeze burns.

Note to the physician: Avoid administering adrenaline or any other similar products as cardiac arrhythmia may result with possible subsequent cardiac arrest.

5) Fire- Fighting Measures

Extinguishing Media: All extinguishing agents can be used.

Specific Hazards: HCFC 22 is not flammable under ambient conditions of temperature and pressure. Certain mixtures of HCFC 22 and air when under pressure may be flammable. Mixtures of HCFC and Air under pressure should be avoided. Certain mixtures of HCFCs and Chlorine may be flammable or reactive under certain conditions. Thermal decomposition will evolve very toxic and corrosive vapours. Containers may burst if over heated.

Specific Fire Fighting Methods: Stay upwind. Evacuate the personnel away from the fumes. Cool down the containers/ equipment exposed to heat with water spray.

Protection of the Fire Fighters: Self Contained breathing apparatus and full protective clothing must be worn in fire conditions.

6) Accidental Release Measures

Personal Precautions: Avoid contact with skin and eyes, Do not breathe gas. No naked Flames. Do not smoke. For further information refer to section 8 "exposure controls/personal protection".

Heavy vapours, shut off low level openings in the vicinity(ventilation shafts, drains) prevent the product from entering cellars, basements, or pits since the vapour may create a suffocating atmosphere. Provided it is safe to do so, isolate the source of the leak. Allow small spillages to evaporate provided there is adequate ventilation.

Large Spillages: Ventilate spillage area, contain spillage with earth or sand or any suitable absorbent material.

Environmental Precautions: Contain the spilled material, prevent the product from spreading into the environment.

Methods of cleaning up: Recover as much product as possible, allow residual product to evaporate for disposal of contaminated material refer to section 13.

7) Handling and Storage

Handling

Technical Measures: Ventilation is required, Use Closed systems. Avoid contact with hot surfaces. Avoid High temperatures. Smoking is forbidden. Avoid inhalation of high concentrations of vapours.

Atmospheric levels should be controlled in compliance with occupational exposure limit. The vapour is heavier than air, high concentrations may be produced at low levels where general ventilation is poor, in such cases provide adequate ventilation or wear suitable respiratory protective equipment with positive air supply.

Avoid contact with naked flames as corrosive and very toxic decomposition products can be formed. Avoid contact between the liquid and skin and eyes.

Liquid transfer of refrigerants between refrigerant containers and to and from systems can result in static generation, ensure adequate earthing to avoid this.

Storage

Technical measures: storage area should be equipped with ventilation at low level, Take necessary measures to avoid the accidental release of the product outside, due to rupture of containers or transfer systems.

Storage conditions: Keep the containers tightly closed and dry in a cool and well ventilated area at temperatures not exceeding 45C and away from any source of heat including direct sun light. Avoid storing near to the intake of air conditioning units, boiler units or open drains and away from any source of Ignition.

Packaging material: Steel is recommended.

8) Exposure Controls/ Personal Protection

Engineering measures: Ensure good ventilation of the work station

Occupational exposure limits: Exposure limits LTEL 1,000ppm 8 hr TWA 3590mg/m³
OES

Personal Protective Equipment: In event of insufficient ventilation self contained apparatus is required, Refrigerants should only be handled with the use of protective gloves insulated against the cold. Eyes should be protected with goggles and to protect the skin and body Impermeable clothing should be worn. Do not drink eat or smoke in the workplace.

9) Physical and Chemical Properties

Appearance

Physical State is compressed Liquefied gas, colourless with a slight ethereal odour.

Specific temperatures: Boiling Point 40.8°C, melting point 160°C

Oxidising properties: Non oxidising material according to EEC criteria

Vapour Pressure (mm Hg): 6805 at 20°C

Vapour density: (Air=1)>3.03

Specific Gravity: 1.21

Liquid density: 11.924kg/m³

Solubility in water: 0.30% wt/wt at 25°C

10) Stability and Reactivity

Stability: Stable at ambient temperature and under normal conditions of use

Conditions to Avoid: May decompose on contact with hot surfaces and flames incompatible with Alkali metals, sodium, Potassium, Barium, Magnesium, alloys and powdered metals.

Hazardous decomposition products: On Combustion or thermal decomposition (Pyrolysis) and Hydrolysis releases toxic gasses (halogenated compounds) (Hydrogen Chloride and Hydrogen Fluoride)

11) Toxicological Information

Acute Toxicity: Vapours: Published data

Acute Symptoms: Effects following high level exposure: Headaches, Dizziness, Loss of Consciousness

Inhalation: High exposures may cause abnormal heart rhythm and prove suddenly fatal. Very high atmospheric concentrations may cause anaesthetic effects and asphyxiation,

Skin Contact & Eye Contact: Liquid Splashes or spray may cause freeze burns unlikely to be hazardous through skin absorption.

Ingestion: Highly unlikely but should this occur freeze burns will result

Long terms exposure: A lifetime of inhalation study in animals has shown that high exposure if HCFC 22 (50,000PPM) produce a small excess of salivary gland tumours in male rats. Female rats and both sexes of mouse showed no response. The no effect level was 10,000PPM this information does not suggest that HCFC 22 represents a carcinogenic hazard to humans under normal conditions of handling and use. Contact with liquefied gas causes Frostbite and causes injury to the cornea.

12) Ecological Information

Behaviour in the environment: High tonnage material produced in wholly contained systems. High tonnage material used in open systems.

Mobility: Product is Volatile when in aqueous solution

Persistence/Degradability: *Decomposes comparatively rapidly in lower atmosphere(Troposphere). Atmospheric lifetime is 13.3 Years, products of decomposition will be highly dispersed and hence will have low concentration. Is not a Voc under UNECE agreement, Ozone deleting potential (ODP) is 0.055 measured against 1 for CFC 11 as defined by UNEP, substance is controlled under the Montreal protocol (1992 Revision)*

Bioaccumulation: Non Bioaccumulative

Destination of Product: AIR

13) Disposal Considerations

Waste from Residues: Do not allow the product to be released into the environment, Consult the manufacturer or supplier for information regarding recovery and recycling of the product, if recovery is not possible Incinerate at a licensed installation.

Contaminated Packaging: Degas reusable containers. Return to supplier. Disposable containers disposed of at an authorised land fill site.

The users attention is drawn to the possible existence of local regulations regarding disposal.

14) Transport Information

Road and Rail Transport:

Un No. 1018

ERG No 126

Hazchem warning 2 C Non Flammable Gas

Sea Transportation:

IMDG 1018

Class 2.2

Label Non-flammable gas

Air Transportation:

ICAO/IATA code 1018

Class 2.2

Packaging instructions

- Cargo 200

- Passenger 200

Maximum quantity allowed

- Cargo 150 kg

- Passenger 75 kg

15) Regulatory Information

EEC Hazard class Non-flammable gas
Risk phrases R20 Harmful by inhalation
R34 Liquid phase could cause burns
R44 Risk of explosion when heated under

Confinement

Safety phrases S9 Keep container in a well-ventilated place
S15 Keep away from heat
S23 Do not breathe the gas.
S36 Wear suitable protective clothing
S41 In case of fire/explosion do not breathe

Fumes

S51 Use only in well ventilated areas
S56 Do not discharge into the environment.
Dispose to an authorised waste collection point
National legislation None
Refer to SABS 0265 for explanation of the above.

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